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APPLICATION NO.	FILING DATE	FIRST NAMED INVENTOR	ATTORNEY DOCKET NO.	CONFIRMATION NO.
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CHRISTIE, PARKER & HALE, LLP PO BOX 7068 PASADENA, CA 91109-7068			CHIN, BRAD Y	
		ART UNIT		PAPER NUMBER
		1744		

DATE MAILED: 10/28/2005

Please find below and/or attached an Office communication concerning this application or proceeding.

<b>Office Action Summary</b>	Application No.	Applicant(s)
	10/687,847	TREIMAN, MICHAEL T.
	Examiner Brad Y. Chin	Art Unit 1744

-- The MAILING DATE of this communication appears on the cover sheet with the correspondence address --  
Period for Reply

A SHORTENED STATUTORY PERIOD FOR REPLY IS SET TO EXPIRE 3 MONTH(S) OR THIRTY (30) DAYS, WHICHEVER IS LONGER, FROM THE MAILING DATE OF THIS COMMUNICATION.

- Extensions of time may be available under the provisions of 37 CFR 1.136(a). In no event, however, may a reply be timely filed after SIX (6) MONTHS from the mailing date of this communication.
- If NO period for reply is specified above, the maximum statutory period will apply and will expire SIX (6) MONTHS from the mailing date of this communication.
- Failure to reply within the set or extended period for reply will, by statute, cause the application to become ABANDONED (35 U.S.C. § 133). Any reply received by the Office later than three months after the mailing date of this communication, even if timely filed, may reduce any earned patent term adjustment. See 37 CFR 1.704(b).

#### Status

- 1) Responsive to communication(s) filed on 14 September 2005.
- 2a) This action is **FINAL**.                            2b) This action is non-final.
- 3) Since this application is in condition for allowance except for formal matters, prosecution as to the merits is closed in accordance with the practice under *Ex parte Quayle*, 1935 C.D. 11, 453 O.G. 213.

#### Disposition of Claims

- 4) Claim(s) 1-41 is/are pending in the application.
- 4a) Of the above claim(s) \_\_\_\_\_ is/are withdrawn from consideration.
- 5) Claim(s) 26,40 and 41 is/are allowed.
- 6) Claim(s) 1-25 and 27-39 is/are rejected.
- 7) Claim(s) \_\_\_\_\_ is/are objected to.
- 8) Claim(s) \_\_\_\_\_ are subject to restriction and/or election requirement.

#### Application Papers

- 9) The specification is objected to by the Examiner.
- 10) The drawing(s) filed on \_\_\_\_\_ is/are: a) accepted or b) objected to by the Examiner.  
Applicant may not request that any objection to the drawing(s) be held in abeyance. See 37 CFR 1.85(a).  
Replacement drawing sheet(s) including the correction is required if the drawing(s) is objected to. See 37 CFR 1.121(d).
- 11) The oath or declaration is objected to by the Examiner. Note the attached Office Action or form PTO-152.

#### Priority under 35 U.S.C. § 119

- 12) Acknowledgment is made of a claim for foreign priority under 35 U.S.C. § 119(a)-(d) or (f).
- a) All    b) Some \* c) None of:
  1. Certified copies of the priority documents have been received.
  2. Certified copies of the priority documents have been received in Application No. \_\_\_\_\_.
  3. Copies of the certified copies of the priority documents have been received in this National Stage application from the International Bureau (PCT Rule 17.2(a)).

\* See the attached detailed Office action for a list of the certified copies not received.

#### Attachment(s)

- 1) Notice of References Cited (PTO-892)
- 2) Notice of Draftsperson's Patent Drawing Review (PTO-948)
- 3) Information Disclosure Statement(s) (PTO-1449 or PTO/SB/08)  
Paper No(s)/Mail Date \_\_\_\_\_
- 4) Interview Summary (PTO-413)  
Paper No(s)/Mail Date \_\_\_\_\_
- 5) Notice of Informal Patent Application (PTO-152)
- 6) Other: \_\_\_\_\_

DETAILED ACTION

***Claim Objections***

1. Claim 36 is objected to because of the following informalities: Applicant improperly uses the word, "apparatus" in describing claim 1, where Applicant should have used the word, "method" in describing claim 1. Appropriate correction is required.

***Claim Rejections - 35 USC § 102***

The following is a quotation of the appropriate paragraphs of 35 U.S.C. 102 that form the basis for the rejections under this section made in this Office action:

A person shall be entitled to a patent unless –

(e) the invention was described in (1) an application for patent, published under section 122(b), by another filed in the United States before the invention by the applicant for patent or (2) a patent granted on an application for patent by another filed in the United States before the invention by the applicant for patent, except that an international application filed under the treaty defined in section 351(a) shall have the effects for purposes of this subsection of an application filed in the United States only if the international application designated the United States and was published under Article 21(2) of such treaty in the English language.

2. Claims 1, 7-8, and 36 rejected under 35 U.S.C. 102(e) as being anticipated by Sand et. al. [U.S. Patent No. 6,655,401].

Regarding claim 1, Sand et. al. teach a method for diluting a concentrated solution of sterilant for sterilizing instruments or equipment comprising the steps of: providing an eductor (single device 32 for selectively educting one or more chemical fluids for mixing with a motive fluid or eductor 92), the eductor comprising a metering tip (See col. 5, lines 1-7 – metering tip (not shown) for controlling the dilution ratio of the chemical fluid) having a first orifice size (See col. 5, lines 1-7 – metering tip comprises an orifice allowing the chemical fluid to flow from the chemical fluid reservoir to the chemical inlet port), a chemical inlet port (chemical port 94), and a water inlet port (See Fig. 2 – water inlet port on the top of single device 32 at end of motive fluid conduit 36); hooking a container containing concentrated sterilant to the chemical inlet port of the eductor (See col. 5, lines 4-7 – reservoir containing the chemical fluid (not shown)

connected to the chemical inlet port of the eductor by use of a chemical fluid conduit); hooking a water supply source to the water inlet port of the eductor (See col. 2, line 6 – pressurized water supply at water inlet 16; See col. 4, lines 19-20 – chemical fluids diluted to a desired dilution ratio by mixing with a motive fluid of pressurized water), the water supply source comprising a pressure regulating valve for regulating a working pressure of the water supply from a first pressure to a second pressure (See col. 2, line 6 – the motive fluid is a pressurized water supply at water inlet 16; See col. 4, lines 23-27 – the motive fluid is received at an inlet valve 34, depicted as a ball valve, and passed through a motive fluid conduit 36 to the device 32 for mixing a selected chemical fluid, drawn respectively from chemical fluid reservoirs 38, 40 via chemical conduits 42, 44; and lines 39-40 – device 32 includes a single selection control depicted as lever 52 [for activating the ball valve from a first position to a second position], for selecting the mixing, i.e. for regulating the dilution ratio of the chemical/water mixture), and regulating the pressure regulating valve (ball valve/inlet valve 34) based on the first pressure (completely closed or at a first position with a first pressure) to maintain the water supply source at substantially the second pressure (completely open or at an operator-determined second position with a second position) [to produce a regulated working pressure of the water supply for controlling the desired dilution ratio of the chemical fluids with the motive fluid]); activating the eductor by opening a valve to mix water and concentrated sterilant to a desired admixture containing a volume of sterilant to a volume of water (See col. 4, lines 39-40 – single selection control 52 for selecting [activating] the mixing and for simplifying use of the dispenser for the user); adjusting the admixture by varying the volume of the sterilant to the volume of water by varying at least one of the metering tip to one having a second orifice size and the working pressure of the water supply to the eductor (See col. 5, lines 1-7 – typically a metering tip (not shown) is inserted into the chemical port 94 for controlling the dilution ratio of the chemical fluid

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in coordination with the dimensional sizing of the eductor; See col. 2, line 6 – the motive fluid is a pressurized water supply at water inlet 16; See col. 4, lines 23-27 – the motive fluid is received at an inlet valve 34, depicted as a ball valve, and passed through a motive fluid conduit 36 to the device 32 for mixing a selected chemical fluid, drawn respectively from chemical fluid reservoirs 38, 40 via chemical conduits 42, 44. The inlet valve 34, i.e. the ball valve, rotates as inlet valve 34 is rotated between a first position to a second position, functioning to regulate the working pressure of the water supply entering the water inlet port of the eductor); and using the admixture to sterilize an instrument for use in treating a subject (See col. 4, lines 15-22 – a suitable application for dispenser 30 includes dispensing chemical fluids such as a disinfectant, e.g. to sterilize an instrument for use in treating a subject; See col. 6, lines 54-56 – the selected fluids are then dispensed through outlet 48 for uses, such as washing or filling portable dispensing articles, such as spray bottles).

Regarding claim 7, Sand et. al. teach the method, further comprising the steps of providing a second eductor (second eductor 96) and hooking a container containing at least one of a disinfectant (See col. 4, lines 15-22 – one suitable application identifies dispensing chemical fluids such as disinfectants), a sporicide, a biocide, a virucide, or a fungicide to a chemical inlet port (chemical port 98) of the second eductor (second eductor 96).

Regarding claim 8, Sand et. al. teach the method, further comprising a barb assembly connected to the eductor's chemical inlet port and the metering tip is connected to the barb assembly (See Fig. 1 of prior art where metering tip is to be inserted, denoted by dotted lines, into the orifice in the barb assembly attached and projecting laterally from the side of eductors 18 and 20). It is also common knowledge to one of ordinary skill in the art to use metering tips in conjunction with barb assemblies, which are attached to the inlet ports of the eductors, to control the dilution ratio of the chemical fluid(s) entering the eductor.

Regarding claim 36, Sand et. al. teach the method, wherein the concentrated sterilant is one of a concentrated disinfectant, a concentrated antiseptic, a concentrated sporicide, a concentrated biocide, a concentrated virucide, or a concentrated fungicide (See col. 4, lines 15-22 – one suitable application identifies dispensing chemical fluids such as disinfectants. The disinfectant could be characterized as concentrated because it is being diluted with water and potentially another chemical fluid, e.g. a pH-adjusting agent).

#### ***Claim Rejections - 35 USC § 103***

The text of those sections of Title 35, U.S. Code not included in this action can be found in a prior Office action.

The factual inquiries set forth in *Graham v. John Deere Co.*, 383 U.S. 1, 148 USPQ 459 (1966), that are applied for establishing a background for determining obviousness under 35 U.S.C. 103(a) are summarized as follows:

1. Determining the scope and contents of the prior art.
2. Ascertaining the differences between the prior art and the claims at issue.
3. Resolving the level of ordinary skill in the pertinent art.
4. Considering objective evidence present in the application indicating obviousness or nonobviousness.

This application currently names joint inventors. In considering patentability of the claims under 35 U.S.C. 103(a), the examiner presumes that the subject matter of the various claims was commonly owned at the time any inventions covered therein were made absent any evidence to the contrary. Applicant is advised of the obligation under 37 CFR 1.56 to point out the inventor and invention dates of each claim that was not commonly owned at the time a later invention was made in order for the examiner to consider the applicability of 35 U.S.C. 103(c) and potential 35 U.S.C. 102(e), (f) or (g) prior art under 35 U.S.C. 103(a).

3. Claims 31 and 32 are rejected under 35 U.S.C. 103(a) as being unpatentable over Sand et. al.

Regarding claim 31, Sand et. al. teach the method as described above in paragraph 1. Sand et. al. further teach the step of adjusting the metering tip to one having a second orifice (See col. 5, lines 1-7 – typically a metering tip (not shown) is inserted into the chemical port 94. for controlling the dilution ratio of the chemical fluid in coordination with the dimensional sizing of the eductor). Sand et. al. further teach the step of adjusting the working pressure [of the pressurized water supply] up or down into the inlet of the eductor (See col. 2, line 6 – the motive fluid is a pressurized water supply at water inlet 16; See col. 4, lines 23-27 – the motive fluid is received at an inlet valve 34, depicted as a ball valve, and passed through a motive fluid conduit 36 to the device 32 for mixing a selected chemical fluid, drawn respectively from chemical fluid reservoirs 38, 40 via chemical conduits 42, 44. The inlet valve 34, i.e. the ball valve, rotates as inlet valve 34 is rotated between a first position to a second position, functioning to regulate the working pressure of the water supply entering the water inlet port of the eductor). However, Sand et. al. fail to teach the step of adjusting the working pressure up or down while leaving the metering tip with the first orifice size alone to vary the admixture of sterilant and water. It would have been obvious to one of ordinary skill in the art at the time the invention was made to adjust the working pressure of the water, i.e. adjust the amount of water mixed with the concentrated sterilant, while leaving the metering tip with the first orifice alone, i.e. limiting the amount of sterilant into the eductor, because adjusting the water through the eductor while limiting the amount of sterilant entering the eductor allows the user to produce a more dilute sterilant admixture, which may be required for applications sensitive to a concentrated sterilant.

Regarding claim 32, Sand et. al. teach the method as described above in paragraph 1. Sand et. al. further teach the step of adjusting the metering tip to one having a second orifice

(See col. 5, lines 1-7 – typically a metering tip (not shown) is inserted into the chemical port 94 for controlling the dilution ratio of the chemical fluid in coordination with the dimensional sizing of the eductor). Sand et. al. further teach the step of adjusting the working pressure [of the pressurized water supply] up or down into the inlet of the eductor (See col. 2, line 6 – the motive fluid is a pressurized water supply at water inlet 16; See col. 4, lines 23-27 – the motive fluid is received at an inlet valve 34, depicted as a ball valve, and passed through a motive fluid conduit 36 to the device 32 for mixing a selected chemical fluid, drawn respectively from chemical fluid reservoirs 38, 40 via chemical conduits 42, 44. The inlet valve 34, i.e. the ball valve, rotates as inlet valve 34 is rotated between a first position to a second position position, functioning to regulate the working pressure of the water supply entering the water inlet port of the eductor). However, Sand et. al. fail to teach the step of adjusting the metering tip to one having a second orifice, while leaving the working pressure alone to vary the admixture of sterilant and water. It would have been obvious to one of ordinary skill in the art at the time the invention was made to adjust the metering tip to one having a second orifice, i.e. allowing more concentrated sterilant into the eductor, while leaving the working pressure [of the pressurized water supply] alone, i.e. limiting the amount of water into the eductor, because adjusting the amount of concentrated sterilant through the eductor while limiting the amount of water entering the eductor allows the user to maintain a more dilute sterilant admixture, which may be required for applications requiring a more concentrated sterilant.

4. Claims 2, 3, and 9 are rejected under 35 U.S.C. 103(a) as being unpatentable over Sand et. al., as applied above in paragraph 2 to claims 1 and 8, respectively, and further in view of Wachman et. al. [U.S. Patent No. 5,242,323].

Regarding claim 2, Sand et. al. teach the method as described above in paragraph 2, but fail to teach the concentrated sterilant is a 50% or less by weight solution of glutaraldehyde to water. Wachman et. al. teach a sterilant composition wherein the concentrated sterilant is a 50% or less by weight solution of glutaraldehyde to water (See col. 10, lines 51-57 – Example A). It would have been obvious to one of ordinary skill in the art at the time the invention was made to incorporate the sterilant composition of Wachman et. al. into the method of Sand et. al. because Sand et. al. teach the dispensing of chemical fluids, such as disinfectants – the sterilant composition, as taught by Wachman et. al., that are diluted to a desired dilution ratio by mixing with a motive fluid of pressurized water.

Regarding claim 3, Sand et. al. teach the method as described above in paragraph 2, and further teach a second eductor 96 comprising a second chemical inlet [port] 98 and wherein a container containing a second chemical fluid is connected to the second chemical inlet port (See col. 5, lines 8-14). Sand et. al. fail to teach a pH-adjusting agent is connected to the second chemical inlet port. Wachman et. al. teach a sterilant composition that includes a concentrated sterilant, water, and a diluent, such as a pH-adjusting agent (See col. 4, lines 23-33 – “typical embodiment of the present invention comprises: alkylbenzyldimethylammonium chloride, cetyltrimethylammonium bromide, glutaraldehyde, Isopropyl alcohol, propylene glycol, sodium nitrite, tetrasodium ethylenediamine tetraacetate, and water). It would have been obvious to one of ordinary skill in the art at the time the invention was made to incorporate the sterilant composition comprising a pH-adjusting agent, as taught by Wachman et. al., into the method of Sand et. al. because Sand et. al. provides the motivation for controlling the dilution ratio of multiple chemical solutions, where a first eductor produces a diluted admixture from one concentrated sterilant composition and a second eductor which produces a second diluted

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adixture from a second concentrated sterilant composition, such as the sterilant composition comprising a pH adjusting agent, as taught by Wachman et. al.

Regarding claim 9, Sand et. al. teach the method as described above in paragraph 2, but fail to teach the admixture produced comprises about a 3.2% by weight of glutaraldehyde, about a 0.925% by weight of the pH adjusting agent, and a balance by weight of water. Wachman et. al. teach an admixture, comprising about a 3.2% by weight of glutaraldehyde, about a 0.925% by weight of the pH adjusting agent, and a balance by weight of water (See col. 13, Example 5). It would have been obvious to one of ordinary skill in the art at the time the invention was made to incorporate the sterilant composition comprising a pH-adjusting agent with the aforementioned constituent percentages, as taught by Wachman et. al., into the method of Sand et. al. because Sand et. al. provides the motivation for controlling the dilution ratio of multiple chemical solutions, where a first eductor produces a diluted admixture from one concentrated sterilant composition and a second eductor which produces a second diluted admixture from a second concentrated sterilant composition, such as the sterilant composition comprising a pH adjusting agent with the aforementioned constituent percentages, as taught by Wachman et. al.

5. Claims 19, 21, and 23-25 are rejected under 35 U.S.C. 103(a) as being unpatentable over Sand et. al. in view of Wachman et. al.

Regarding claim 19, Sand et. al. teach an apparatus for diluting a concentrate comprising: a proportioning and dispensing unit comprising at least two eductors (dispenser 30 with first eductor 92 and second eductor 96), wherein a first eductor comprises a first chemical inlet port ([first] chemical inlet port 94), a second chemical inlet port ([second] chemical inlet port 98), a motive source inlet port (See Fig. 2 – motive fluid conduit 36 from a pressurized water

supply connected to eductor's water inlet port), and an outlet port (outlet 48); a first container containing a concentrate having a container outlet port and a first hose connecting the container outlet port to the first chemical inlet port (See col. 5, lines 1-7 – reservoir containing the chemical fluid (not shown) connected to the chemical inlet port of the eductor by use of a chemical fluid conduit); a line connecting a motive source to the motive source inlet port (motive fluid conduit 36), the line comprising a regulating valve for pressure regulating pressure supplied by the motive source from a first pressure (valve completely closed or at a first position with a first pressure) to a second pressure (completely open or at an operator-determined second position with a second position) [to produce a regulated working pressure of the water supply for controlling the desired dilution ratio of the chemical fluids with the motive fluid]), which is lower than the first pressure (See col. 2, line 6 – the motive fluid is a pressurized water supply at water inlet 16; See col. 4, lines 23-27 – the motive fluid is received at an inlet valve 34, depicted as a ball valve, and passed through a motive fluid conduit 36 to the device 32 for mixing a selected chemical fluid, drawn respectively from chemical fluid reservoirs 38, 40 via chemical conduits 42, 44; and lines 39-40 – device 32 includes a single selection control depicted as lever 52 [for activating the ball valve from a first position to a second position], for selecting the mixing, i.e. for regulating the dilution ratio of the chemical/water mixture); a third hose (hose 50) for connecting to the outlet port (outlet 48) of the eductor (device 32); a valve to permit regulated motive source from the pressure regulating valve at the second pressure to flow through the first eductor (See col. 2, line 6 – the motive fluid is a pressurized water supply at water inlet 16; See col. 4, lines 23-27 – the motive fluid is received at an inlet valve 34, depicted as a ball valve, and passed through a motive fluid conduit 36 to the device 32 for mixing a selected chemical fluid, drawn respectively from chemical fluid reservoirs 38, 40 via chemical conduits 42, 44; and lines 39-40 – device 32 includes a single selection control depicted as lever 52 [for activating the ball

valve from a first position to a second position], for selecting the mixing, i.e. for regulating the dilution ratio of the chemical/water mixture; See col. 6, lines 46-54 – a dispenser 32 mixes a selected chemical fluid when an operator positions a selection member, such as a selector disk 68, concentric disk selection member 122, push button selection member 152, and a cylindrical section member 202. Thereby, the device 32 diverts motive fluid to one or more fluids channels, such as to the first eductor 92 that draws a first chemical fluid for mixing, to the second eductor 96 that draws a second chemical fluid for mixing, or to the motive fluid bypass 100); and wherein a first metering tip is removably received in the first chemical inlet port and a second metering tip is removably received in the second chemical inlet port (See col. 5, lines 1-13 – metering tips (not shown) inserted [removably received] into the [first] chemical port 94 of first eductor 92 and the [second] chemical port 98 of second eductor 96, respectively). Sand et. al. fail to teach a second container containing a pH adjusting agent having a container outlet port and a second hose connecting the container outlet port to the second chemical inlet port. Sand et. al. identify the use of a second container comprising an outlet port and a second chemical fluid conduit connecting the container to the second chemical inlet port 98 (See Specification, col. 5, lines 8-13); however, Sand et. al. do not specifically identify that the second chemical element is a pH-adjusting agent. Wachman et. al. teach a sterilant composition that includes a concentrated sterilant, water, and a diluent, such as a pH-adjusting agent (See col. 4, lines 23-33 – “typical embodiment of the present invention comprises: alkylbenzyldimethylammonium chloride, cetyltrimethylammonium bromide, glutaraldehyde, isopropyl alcohol, propylene glycol, sodium nitrite, tetrasodium ethylenediamine tetraacetate, and water). It would have been obvious to one of ordinary skill in the art at the time the invention was made to incorporate the sterilant composition comprising a pH-adjusting agent, as taught by Wachman et. al., into the method of Sand et. al. because Sand et. al. provides the

motivation for controlling the dilution ratio of multiple chemical solutions, where a first eductor produces a diluted admixture from one concentrated sterilant composition and a second eductor which produces a second diluted admixture from a second concentrated sterilant composition, such as the sterilant composition comprising a pH adjusting agent, as taught by Wachman et. al.

Regarding claim 21, Sand et. al. and Wachman et. al. teach the apparatus as described above. Wachman et. al. further teach a sterilant composition wherein the concentrated sterilant is a 50% or less by weight solution of glutaraldehyde to water (See col. 10, lines 51-57 – Example A). It would have been obvious to one of ordinary skill in the art at the time the invention was made to incorporate the sterilant composition of Wachman et. al. into the method of Sand et. al. because Sand et. al. teach the dispensing of chemical fluids, such as disinfectants – the sterilant composition, as taught by Wachman et. al., that are diluted to a desired dilution ratio by mixing with a motive fluid of pressurized water.

Regarding claim 23, Sand et. al. teach the apparatus, further comprising a barb assembly connected to the eductor's chemical inlet port and the metering tip is connected to the barb assembly (See Fig. 1 of prior art where metering tip is to be inserted, denoted by dotted lines, into the orifice in the barb assembly attached and projecting laterally from the side of eductors 18 and 20). It is also common knowledge to one of ordinary skill in the art to use metering tips in conjunction with barb assemblies, which are attached to the inlet ports of the eductors, to control the dilution ratio of the chemical fluid(s) entering the eductor).

Regarding claim 24, Sand et. al. teach the apparatus, wherein the third hose is directed to a holding container for outputting the admixture into the holding container (hose 50; See col. 6, lines 54-56 – the selected fluids are then dispensed through an outlet 48 for uses, such as for filling portable dispensing articles, such as spray bottles).

Regarding claim 25, Sand et. al. teach the apparatus, further comprising a container containing at least one of a disinfectant (See col. 4, lines 15-22 – one suitable application identifies dispensing chemical fluids such as disinfectants), a sporicide, a biocide, a virucide, or a fungicide to a chemical inlet port (chemical port 98) of the second eductor (second eductor 96).

6. Claim 5 is rejected under 35 U.S.C. 103(a) as being unpatentable over Sand et. al., as applied above in paragraph 2, and further in view of Marais et. al. [U.S. Patent Publication No. 2004/0037737].

Regarding claim 5, Sand et. al. teach the method as described above in paragraph 2, but fail to teach the step of mounting the eductor in a health care facility. Marais et. al. teach a method of and equipment for washing, disinfecting, and/or sterilizing health care devices, including medical, dental, or veterinary equipment, as well as cooking and catering materials (See page 1, [0001]) with an aqueous solution that is electrochemically activated. It would have been obvious to one of ordinary skill in the art at the time the invention was made to combine the teachings of Sand et. al. and Marais et. al. and to mount an eductor in a health care facility in order to produce an adequate sterilant and water admixture for use in sterilizing, washing, and/or disinfecting health care devices, including medical, dental, or veterinary equipment in a health care facility by the process and method, as taught by Marais et. al.

7. Claims 10, 12, 13, 16, and 17 are rejected under 35 U.S.C. 103(a) as being unpatentable over Sand et. al. in view of Marais et. al.

Regarding claim 10, Sand et. al. teach a method for diluting a concentrated chemical solution with water, comprising the steps of: providing an eductor housed in a housing (single

device [eductor] 32 or eductor 92 in dispenser 30); adjusting the eductor's output by adjusting a pressure regulating valve to adjust a water supply pressure from a first pressure to a second pressure and regulating the pressure regulating valve to maintain the water supply pressure at substantially the second pressure (See col. 2, line 6 – the motive fluid is a pressurized water supply at water inlet 16; See col. 4, lines 23-27 – the motive fluid is received at an inlet valve 34, depicted as a ball valve, and passed through a motive fluid conduit 36 to the device 32 for mixing a selected chemical fluid, drawn respectively from chemical fluid reservoirs 38, 40 via chemical conduits 42, 44; and lines 39-40 – device 32 includes a single selection control depicted as lever 52 [for activating the ball valve from a first position to a second position], for selecting the mixing, i.e. for regulating the dilution ratio of the chemical/water mixture) and adjusting a chemical inlet back pressure by selecting a metering tip having a first orifice size (See col. 5, lines 1-7 – metering tip (not shown) with an orifice inserted into the chemical port 94 for controlling the dilution ratio of the chemical fluid); hooking an inlet connected to a container containing the concentrated chemical solution to the eductor's chemical inlet port (See col. 5, lines 4-7 – reservoir containing the chemical fluid (not shown) connected to the chemical inlet port of the eductor by use of a chemical fluid conduit); hooking an inlet line from a water supply source downstream of the regulating valve to the eductor's water inlet port (See Fig. 2 – motive fluid conduit 36 from a pressurized water supply connected to eductor's water inlet port); activating the eductor by opening a valve so that regulated water flows through the water inlet port and concentrated chemical solution flows through the chemical inlet port (See col. 4, lines 39-40 – device 32 includes a single selection control depicted as lever 52 [for activating the ball valve from a first position to a second position], for selecting the mixing, i.e. for regulating the dilution ratio of the chemical/water mixture; See col. 6, lines 46-54 – a dispenser 32 mixes a selected chemical fluid when an operator positions a selection member, such as a selector disk

68, concentric disk selection member 122, push button selection member 152, and a cylindrical section member 202. Thereby, the device 32 diverts motive fluid to one or more fluids channels, such as to the first eductor 92 that draws a first chemical fluid for mixing, to the second eductor 96 that draws a second chemical fluid for mixing, or to the motive fluid bypass 100); outputting the admixture into a holding container (See col. 6, lines 54-56 – the selected fluids are then dispensed through an outlet 48 for uses, such as for filling portable dispensing articles, such as spray bottles); and applying the mixture to a surface (See col. 6, lines 54-56 – the selected fluids are then dispensed through an outlet 48 for uses, such as washing portable dispensing articles). Sand et. al. fail to particularly teach that the application of the admixture to a surface is inside a health care facility. Marais et. al. teach a method of and equipment for washing, disinfecting, and/or sterilizing health care devices, including medical, dental, or veterinary equipment, as well as cooking and catering materials (See page 1, [0001]) with an aqueous solution that is electrochemically activated. It would have been obvious to one of ordinary skill in the art at the time the invention was made to combine the teachings of Sand et. al. and Marais et. al. because Sand et. al. teaches the application of the admixture for uses, such as washing portable dispensing articles, e.g. spray bottles, where the method of Sand et. al. could be used to apply the admixture to the surfaces of a spray bottle or other medical equipment found in a health care facility, as taught by Marais et. al.

Regarding claim 12, Sand et. al. further teach the eductor is equipped with a second metering tip comprising a second orifice size (See col. 5, lines 1-14).

Regarding claim 13, Sand et. al. teach the eductor comprises a second chemical inlet [port] 98 and wherein a container containing a second chemical fluid is connected to the second chemical inlet port (See col. 5, lines 8-14).

Regarding claim 16, Sand et. al. teach the step of providing a second eductor (second eductor 96) and hooking a container containing at least one of a disinfectant (See col. 4, lines 15-22 – one suitable application identifies dispensing chemical fluids such as disinfectants), a sporicide, a biocide, a virucide, or a fungicide to a chemical inlet port (chemical port 98) of the second eductor (second eductor 96).

Regarding claim 17, Sand et. al. teach the method, further comprising a barb assembly connected to the eductor's chemical inlet port and the metering tip is connected to the barb assembly (See Fig. 1 of prior art where metering tip is to be inserted, denoted by dotted lines, into the orifice in the barb assembly attached and projecting laterally from the side of eductors 18 and 20). It is also common knowledge to one of ordinary skill in the art to use metering tips in conjunction with barb assemblies, which are attached to the inlet ports of the eductors, to control the dilution ratio of the chemical fluid(s) entering the eductor.

8. Claim 6 is rejected under 35 U.S.C. 103(a) as being unpatentable over Sand et. al., as applied above in paragraph 2, and further in view of Grune et. al. [U.S. Patent No. 6,293,153].

Sand et. al. teach the method as described above in paragraph 1, but fail to teach the step of providing a pressure gauge downstream of the regulating valve. Grune et. al. teach a pressure gauge positioned downstream of a regulating valve (pressure regulator 10) for measuring the pressure of the fluidizing gas being admitted into the system (See col. 2, lines 40-45). It would have been obvious to one of ordinary skill in the art at the time the invention was made to incorporate a step for providing a pressure gauge, as taught by Grune et. al., into the method of Sand et. al. because a pressure gauge would function to measure the working pressure of the pressurized water supply entering the eductor inlet.

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9. Claims 11 and 18 are rejected under 35 U.S.C. 103(a) as being unpatentable over Sand et. al. in view of Marais et. al., as applied above in paragraph 7, and further in view of Wachman et. al.

Regarding claim 11, Sand et. al. and Marais et. al. teach the method as described above in paragraph 7, but fail to teach the concentrated sterilant is a 50% or less by weight solution of glutaraldehyde to water. Wachman et. al. teach a sterilant composition wherein the concentrated sterilant is a 50% or less by weight solution of glutaraldehyde to water (See col. 10, lines 51-57 – Example A). It would have been obvious to one of ordinary skill in the art at the time the invention was made to incorporate the sterilant composition of Wachman et. al. into the methods of Sand et. al. and Marais et. al. because Sand et. al. teach the dispensing of chemical fluids, such as disinfectants – the sterilant composition, as taught by Wachman et. al., that are diluted to a desired dilution ratio by mixing with a motive fluid of pressurized water.

Regarding claim 18, Sand et. al. and Marais et. al. teach the method as described above in paragraph 7, but fail to teach the admixture produced comprises about a 3.2% by weight of glutaraldehyde, about a 0.925% by weight of the pH adjusting agent, and a balance by weight of water. Wachman et. al. teach an admixture, comprising about a 3.2% by weight of glutaraldehyde, about a 0.925% by weight of the pH adjusting agent, and a balance by weight of water (See col. 13, Example 5). It would have been obvious to one of ordinary skill in the art at the time the invention was made to incorporate the sterilant composition comprising a pH-adjusting agent with the aforementioned constituent percentages, as taught by Wachman et. al., into the methods of Sand et. al. and Marais et. al. because Sand et. al. provides the motivation for controlling the dilution ratio of multiple chemical solutions, where a first eductor produces a diluted admixture from one concentrated sterilant composition and a second eductor which produces a second diluted admixture from a second concentrated sterilant composition, such as

the sterilant composition comprising a pH adjusting agent with the aforementioned constituent percentages, as taught by Wachman et. al.

10. Claim 15 is rejected under 35 U.S.C. 103(a) as being unpatentable over Sand et. al. in view of Marais et. al., as applied above in paragraph 7, and further in view of Grune et. al.

Sand et. al. and Marais et. al. teach the method as described above in paragraph 7, but fail to teach the step of providing a pressure gauge downstream of the regulating valve. Grune et. al. teach a pressure gauge positioned downstream of a regulating valve (pressure regulator 10) for measuring the pressure of the fluidizing gas being admitted into the system (See col. 2, lines 40-45). It would have been obvious to one of ordinary skill in the art at the time the invention was made to incorporate a step for providing a pressure gauge, as taught by Grune et. al., into the methods of Sand et. al. and Marais et. al. because a pressure gauge would function to measure the working pressure of the pressurized water supply entering the eductor inlet.

11. Claim 22 is rejected under 35 U.S.C. 103(a) as being unpatentable over Sand et. al. in view of Wachman et. al., as applied above in paragraph 5, and further in view of Bristor [U.S. Patent Publication No. 2003/0150936].

Sand et. al. and Wachman et. al. teach the method as described above in paragraph 5, but fail to teach the first metering tip and the second metering tip each comprise an orifice. Bristor teaches that since the size of the orifice 108 of the metering tip 106 controls the amount of chemical drawn and thus controls the dilution ratio, the user need only select the proper metering tip 106 to match the desired dilution ration and thread it into the threaded insert 102 (See page 6, [0086]). It would have been obvious to one of ordinary skill in the art at the time the invention was made to incorporate the metering tips comprising an orifice, as taught by

Bristor into the apparatus of Sand et. al. and Wachman et. al. because choosing an appropriate metering tip comprising an orifice provides allows a user to control the desired dilution ratio.

12. Claim 27 is rejected under 35 U.S.C. 103(a) as being unpatentable over Sand et. al. in view of Marais et. al. and Bristor.

Regarding claim 27, Sand et. al. teach a method for dispensing an admixture of concentrated chemical solution and water in a proportioning and dispensing unit comprising: mounting two eductors to a housing and mounting (first eductor 92 and second eductor 96 mounted in dispenser 30), the two eductors having a common water inlet header (motive fluid passage 120); connecting a first chemical to a chemical inlet port of the first eductor (See col. 5, lines 1-7); connecting a second chemical to a chemical inlet port of the second eductor (See col. 5, lines 8-14); connecting a water supply line to the common water inlet header (See col. 4, line 23); the water supply line comprising a pressure regulating valve to regulate water pressure from a first pressure to a second pressure (See col. 2, line 6 – the motive fluid is a pressurized water supply at water inlet 16; See col. 4, lines 23-27 – the motive fluid is received at an inlet valve 34, depicted as a ball valve, and passed through a motive fluid conduit 36 to the device 32 for mixing a selected chemical fluid, drawn respectively from chemical fluid reservoirs 38, 40 via chemical conduits 42, 44; and lines 39-40 – device 32 includes a single selection control depicted as lever 52 [for activating the ball valve from a first position to a second position], for selecting the mixing, i.e. for regulating the dilution ratio of the chemical/water mixture) and adjusting a chemical inlet back pressure by selecting a metering tip having a first orifice size (See col. 5, lines 1-7 – metering tip (not shown) with an orifice inserted into the chemical port 94 for controlling the dilution ratio of the chemical fluid); activating at least one of the first eductor or the second eductor by opening a valve to open a port on the activated eductor to produce an

admixture of at least one of the first chemical and water or the second chemical and water (See col. 4, lines 39-40 – device 32 includes a single selection control depicted as lever 52 [for activating the ball valve from a first position to a second position], for selecting the mixing, i.e. for regulating the dilution ratio of the chemical/water mixture; See col. 6, lines 46-54 – a dispenser 32 mixes a selected chemical fluid when an operator positions a selection member, such as a selector disk 68, concentric disk selection member 122, push button selection member 152, and a cylindrical section member 202. Thereby, the device 32 diverts motive fluid to one or more fluids channels, such as to the first eductor 92 that draws a first chemical fluid for mixing, to the second eductor 96 that draws a second chemical fluid for mixing, or to the motive fluid bypass 100); and wherein the chemical inlets of the first and second eductors each comprises a metering tip having an orifice (metering tips (not shown) – See col. 4, lines 1-14). Sand et. al. fail to teach that the metering tips corresponding to the first and second eductor have orifices. Further Sand et. al. fail to teach . Bristor teaches that since the size of the orifice 108 of the metering tip 106 controls the amount of chemical drawn and thus controls the dilution ratio, the user need only select the proper metering tip 106 to match the desired dilution ration and thread it into the tressed insert 102 (See page 6, [0086]). Sand et. al. fail to particularly teach that the application of the admixture to a surface is inside a health care facility. Marais et. al. teach a method of and equipment for washing, disinfecting, and/or sterilizing health care devices, including medical, dental, or veterinary equipment, as well as cooking and catering materials (See page 1, [0001]) with an aqueous solution that is electrochemically activated. It would have been obvious to one of ordinary skill in the art at the time the invention was made to combine the teachings of Sand et. al., Marais et. al., and Wachman et. al. because Sand et. al. teaches the application of the admixture for uses, such as washing portable dispensing articles, e.g. spray bottles, where the method of Sand et. al. could be used with an eductor mounted to a

health care facility to produce and apply an adequate sterilant and water apply the admixture for use in sterilizing, washing, and/or disinfecting health care devices, including medical, dental, or veterinary equipment in a health care facility by the process and method, as taught by Marais et. al. Further, the teachings of Bristor enhance the method of Sand et. al. in providing metering tips comprising an orifice, allowing a user to control the desired dilution ratio of the chemical solution/water admixture.

13. Claims 28-30 are rejected under 35 U.S.C. 103(a) as being unpatentable over Sand et. al. in view of Marais et. al. and Bristor, as applied above in paragraph 12, and further in view of Wachman et. al.

Regarding claim 28, Sand et. al., Marais et. al. and Bristor teach the method as described above in paragraph 12, but fail to teach the concentrated sterilant is a 50% or less by weight solution of glutaraldehyde to water. Wachman et. al. teach a sterilant composition wherein the concentrated sterilant is a 50% or less by weight solution of glutaraldehyde to water (See col. 10, lines 51-57 – Example A). It would have been obvious to one of ordinary skill in the art at the time the invention was made to incorporate the sterilant composition of Wachman et. al. into the method of Sand et. al., Marais et. al. and Bristor because Sand et. al. teach the dispensing of chemical fluids, such as disinfectants – the sterilant composition, as taught by Wachman et. al., that are diluted to a desired dilution ratio by mixing with a motive fluid of pressurized water.

Regarding claim 29, Sand et. al., Marais et. al. and Bristor teach the method as described above in paragraph 12, but fail to teach the method, wherein the second chemical is a pH-adjusting agent. Wachman et. al. teach a sterilant composition that includes a concentrated sterilant, water, and a diluent, such as a pH-adjusting agent (See col. 4, lines 23-33 – “typical

embodiment of the present invention comprises: alkylbenzyldimethylammonium chloride, cetyltrimethylammonium bromide, glutaraldehyde, Isopropyl alcohol, propylene glycol, sodium nitrite, tetrasodium ethylenediamine tetraacetate, and water). It would have been obvious to one of ordinary skill in the art at the time the invention was made to incorporate the sterilant composition comprising a pH-adjusting agent, as taught by Wachman et. al., into the method of Sand et. al. because Sand et. al. provides the motivation for controlling the dilution ratio of multiple chemical solutions, where a first eductor produces a diluted admixture from one concentrated sterilant composition and a second eductor which produces a second diluted admixture from a second concentrated sterilant composition, such as the sterilant composition comprising a pH adjusting agent, as taught by Wachman et. al.

Regarding claim 30, Sand et. al., Marais et. al. and Bristor teach the method as described above in paragraph 12, but fail to teach the method, wherein the first chemical is a 50% by weight of glutaraldehyde to water and the second chemical is a diluent. Wachman et. al. teach a sterilant composition wherein the concentrated sterilant is a 50% or less by weight solution of glutaraldehyde to water (See col. 10, lines 51-57 – Example A). It would have been obvious to one of ordinary skill in the art at the time the invention was made to incorporate the sterilant composition of Wachman et. al. into the method of Sand et. al., Marais et. al. and Bristor because Sand et. al. teach the dispensing of chemical fluids, such as disinfectants – the sterilant composition, as taught by Wachman et. al., that are diluted to a desired dilution ratio by mixing with a motive fluid of pressurized water. Wachman et. al. further teach a sterilant composition that includes a concentrated sterilant, water, and a diluent, such as a pH-adjusting agent (See col. 4, lines 23-33 – “typical embodiment of the present invention comprises: alkylbenzyldimethylammonium chloride, cetyltrimethylammonium bromide, glutaraldehyde, Isopropyl alcohol, propylene glycol, sodium nitrite, tetrasodium ethylenediamine tetraacetate,

and water). It would have been obvious to one of ordinary skill in the art at the time the invention was made to incorporate a first sterilant composition wherein the concentrated sterilant is a 50% or less by weight solution of glutaraldehyde to water and a diluent, both as taught by Wachman et. al., into the process of Sand et. al. because Sand et. al. provides the motivation for controlling the dilution ratio of multiple chemical solutions, where a first eductor produces a diluted admixture from one concentrated sterilant composition and a second eductor which produces a second diluted admixture from a second concentrated sterilant composition.

14. Claim 4 is rejected under 35 U.S.C. 103(a) as being unpatentable over Sand et. al., as applied above in paragraph 2, and further in view of Taylor [U.S. Patent Publication No. 2002/0061263].

Sand et. al. teach the method as described above in paragraph 2, but fail to teach the method, wherein the regulating valve regulates the water supply source to a water pressure of less than 50 psi. Taylor teaches a method for chlorine dioxide generation, comprising regulating the water supply source to a water pressure of 30 psi (See page 4, [0041]). At the time the invention was made, it would have been obvious to a person of ordinary skill in the art to regulate the water supply source to a water pressure of less than 50 psi because Applicant has not disclosed that a water pressure of less than 50 psi provides an advantage, is used for a particular purpose, or solves a stated problem. One of ordinary skill in the art, furthermore, would have expected Applicant's invention to perform equally well at other water pressures. Therefore, in view of Taylor, it would have been obvious to one of ordinary skill in the art to modify Sand et. al. to teach the method of regulating the water supply source to a water pressure less than 50 psi, as specified in claim 4.

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15. Claim 14 is rejected under 35 U.S.C. 103(a) as being unpatentable over Sand et. al. in view of Marais et. al., as applied above in paragraph 7, and further in view of Taylor.

Sand et. al. and Marais et. al. teach the apparatus as described above in paragraph 7, but fail to teach the method, wherein the regulating valve regulates the water supply source to a water pressure of less than 50 psi. Taylor teaches an apparatus for chlorine dioxide generation, comprising regulating the water supply source to a water pressure of 30 psi (See page 4, [0041]). At the time the invention was made, it would have been obvious to a person of ordinary skill in the art to regulate the water supply source to a water pressure of less than 50 psi because Applicant has not disclosed that a water pressure of less than 50 psi provides an advantage, is used for a particular purpose, or solves a stated problem. One of ordinary skill in the art, furthermore, would have expected Applicant's invention to perform equally well at other water pressures. Therefore, in view of Taylor, it would have been obvious to one of ordinary skill in the art to modify Sand et. al. and Marais et. al. to teach the apparatus of regulating the water supply source to a water pressure less than 50 psi, as specified in claim 14.

16. Claim 20 is rejected under 35 U.S.C. 103(a) as being unpatentable over Sand et. al. in view of Wachman et. al., as applied above in paragraph 5, and further in view of Taylor.

Sand et. al., and Wachman et. al. teach the method as described above in paragraph 5, but fail to teach the method, wherein the regulating valve regulates the water supply source to a water pressure of less than 50 psi. Taylor teaches an apparatus for chlorine dioxide generation, comprising regulating the water supply source to a water pressure of 30 psi (See page 4, [0041]). At the time the invention was made, it would have been obvious to a person of ordinary skill in the art to regulate the water supply source to a water pressure of less than 50 psi because Applicant has not disclosed that a water pressure of less than 50 psi provides an

advantage, is used for a particular purpose, or solves a stated problem. One of ordinary skill in the art, furthermore, would have expected Applicant's invention to perform equally well at other water pressures. Therefore, in view of Taylor, it would have been obvious to one of ordinary skill in the art to modify Sand et. al. and Marais et. al. to teach the apparatus of regulating the water supply source to a water pressure less than 50 psi, as specified in claim 20.

17. Claim 33 is rejected under 35 U.S.C. 103(a) as being unpatentable over Sand et. al., as applied above in paragraph 2, and further in view of Stanley [U.S. Patent Publication No. 2004/0156744].

Sand et. al. teach the method as described above in paragraph 2, but fail to teach the step of hooking the water supply source to an inlet of a booster pump and hooking a pump outlet to the water inlet port of the eductor. Stanley teaches a cleaning and sterilizing device and method that utilizes a positive pressure pump to produce positive pressure flow, which provides a greater range of pressures. Stanley further teaches that positive pressure flow provides both the pressure source and the fluid source at one location, simplifying connections and automation. Positive pressure is also safer because contaminants cannot be drawn into the device through leaks in the system (See Specification, p. 4, [0056]). It would have been obvious to one of ordinary skill in the art at the time the invention was made to incorporate the teachings of Stanley into Sand et. al. for providing a pump connected between the motive source and the regulating valve because as Stanley teaches the positive pressure flow would provide the user with a greater range of working pressures for the motive fluid (See page 4, [0056]). Sand et. al. provide the motivation for the control and regulating of the motive fluid pressure into the eductor, providing more control over the amount of motive fluid added in diluting the concentrated chemical fluid. The booster pump creates such a control means for regulating the pressure of

the motive fluid into the eductor and allowing the user to control the dilution ratio of the diluted concentrated solution of sterilant.

18. Claim 34 is rejected under 35 U.S.C. 103(a) as being unpatentable over Sand et. al. in view of Marais et. al., as applied above in paragraph 7, and further in view of Stanley.

Sand et. al. and Wachman et. al. teach the apparatus as described above in paragraph 7, but fail to teach the step of hooking the water supply source to an inlet of a booster pump and hooking a pump outlet to the regulating valve. Stanley teaches a cleaning and sterilizing device and method that utilizes a positive pressure pump to produce positive pressure flow, which provides a greater range of pressures. Stanley further teaches that positive pressure flow provides both the pressure source and the fluid source at one location, simplifying connections and automation. Positive pressure is also safer because contaminants cannot be drawn into the device through leaks in the system (See Specification, p. 4, [0056]). It would have been obvious to one of ordinary skill in the art at the time the invention was made to incorporate the teachings of Stanley into Sand et. al. and Marais et. al. for providing a pump connected between the motive source and the regulating valve because as Stanley teaches the positive pressure flow would provide the user with a greater range of working pressures for the motive fluid (See page 4, [0056]). Sand et. al. provide the motivation for the control and regulating of the motive fluid pressure into the eductor, providing more control over the amount of motive fluid added in diluting the concentrated chemical fluid. The booster pump creates such a control means for regulating the pressure of the motive fluid into the eductor and allowing the user to control the dilution ratio of the diluted concentrated solution of sterilant.

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19. Claim 35 is rejected under 35 U.S.C. 103(a) as being unpatentable over Sand et. al. and Wachman et. al., as applied above in paragraph 5, and further in view of Stanley.

Sand et. al., and Wachman et. al. teach the method as described above in paragraph 5, but fail to teach the step of hooking the water supply source to an inlet of a booster pump and hooking a pump outlet to the regulating valve. Stanley teaches a cleaning and sterilizing device and method that utilizes a positive pressure pump to produce positive pressure flow, which provides a greater range of pressures. Stanley further teaches that positive pressure flow provides both the pressure source and the fluid source at one location, simplifying connections and automation. Positive pressure is also safer because contaminants cannot be drawn into the device through leaks in the system (See Specification, p. 4, [0056]). It would have been obvious to one of ordinary skill in the art at the time the invention was made to incorporate the teachings of Stanley into Sand et. al., and Wachman et. al. for providing a pump connected between the motive source and the regulating valve because as Stanley teaches the positive pressure flow would provide the user with a greater range of working pressures for the motive fluid (See page 4, [0056]). Sand et. al. provide the motivation for the control and regulating of the motive fluid pressure into the eductor, providing more control over the amount of motive fluid added in diluting the concentrated chemical fluid. The booster pump creates such a control means for regulating the pressure of the motive fluid into the eductor and allowing the user to control the dilution ratio of the diluted concentrated solution of sterilant.

20. Claim 37 is rejected under 35 U.S.C. 103(a) as being unpatentable over Sand et. al. in view of Marais et. al., as applied above in paragraph 7, and further in view of Buongiorno et. al. [U.S. Patent Publication No. 2002/0061474].

Sand et. al. and Marais et. al. teach the method as described above in paragraph 7, but fail to teach the concentrated chemical solution is a concentrated photochemical. Buongiorno et. al. teach the use of a concentrated aqueous color developing composition for providing a color image comprising color developing an imagewise exposed color silver halide photographic element. It would have been obvious to one of ordinary skill in the art at the time the invention was made to combine the teachings of Buongiorno et. al. with the methods of Sand et. al. and Marais et. al. because Sand teaches the use of concentrated chemical fluids, such as the concentrated aqueous color developing composition of Buongiorno et. al.

21. Claim 38 is rejected under 35 U.S.C. 103(a) as being unpatentable over Sand et. al. in view of Wachman et. al., as applied above in paragraph 5, and further in view of Buongiorno et. al.

Sand et. al. and Wachman et. al. teach the apparatus as described above in paragraph 5, but fail to teach the concentrated chemical solution is a concentrated photochemical. Buongiorno et. al. teach the use of a concentrated aqueous color developing composition for providing a color image comprising color developing an imagewise exposed color silver halide photographic element. It would have been obvious to one of ordinary skill in the art at the time the invention was made to combine the teachings of Buongiorno et. al. with the apparatus of Sand et. al. and Wachman et. al. because Sand teaches the use of concentrated chemical fluids, such as the concentrated aqueous color developing composition of Buongiorno et. al.

22. Claim 39 is rejected under 35 U.S.C. 103(a) as being unpatentable over Sand et. al. in view of Marais et. al., and Bristor, as applied above in paragraph 12, and further in view of Buongiorno et. al.

Sand et. al., Marais et. al., and Bristor teach the method as described above in paragraph 12, but fail to teach the concentrated chemical solution is a concentrated photochemical. Buongiorno et. al. teach the use of a concentrated aqueous color developing composition for providing a color image comprising color developing an imagewise exposed color silver halide photographic element. It would have been obvious to one of ordinary skill in the art at the time the invention was made to combine the teachings of Buongiorno et. al. with the methods of Sand et. al., Marais et. al., and Bristor because Sand teaches the use of concentrated chemical fluids, such as the concentrated aqueous color developing composition of Buongiorno et. al.

***Allowable Subject Matter***

23. Claims 26 and 40-41 are allowed.

The following is a statement of reasons for the indication of allowable subject matter: Claim 26 includes the limitations for a method, comprising the steps of selecting a hose with unit degradations along at least a portion of the hose length, filling the hose length with a quantity of fluid to a starting fluid level, activating the eductor to produce the admixture of fluid and water at the outlet port, de-activating the eductor, determining an amount of fluid dispensed from the eductor by measuring the unit gradations on the first hose between a starting fluid level and a second fluid level measured after the eductor is de-activated, determining a percent ratio of fluid dispensed to water used to dispense the fluid through the eductor, and if the percent ratio of fluid to water is not as desired, changing at least one of the first metering tip having the first orifice size to a second metering tip having a second orifice size and the first water pressure set point of the pressure regulator to a second water pressure set point, and repeating the steps as necessary until a percent ratio of fluid to water is as desired. None of the references identified

above teach the aforementioned claimed limitations of claim 26 nor would it have been obvious to combine references to achieve the claimed inventive subject matter.

Claims 40-41 include the limitations for an apparatus for diluting a concentrate comprising: a pressure regulating valve for regulating pressure supplied by the motive source from a first pressure to a second pressure, which is lower than the first pressure, and a block valve for at least one of opening and blocking a motive source connected in series with the pressure regulating valve, e.g. two separate valves. Sand et. al. teach a pressure regulating valve for regulating pressure supplied by the motive source from a first pressure to a second pressure, which is lower than the first pressure (inlet ball valve 34 with lever 52; See col. 2, line 6 – the motive fluid is a pressurized water supply at water inlet 16; See col. 4, lines 23-27 – the motive fluid is received at an inlet valve 34, depicted as a ball valve, and passed through a motive fluid conduit 36 to the device 32 for mixing a selected chemical fluid, drawn respectively from chemical fluid reservoirs 38, 40 via chemical conduits 42, 44; and lines 39-40 – device 32 includes a single selection control depicted as lever 52 [for activating the ball valve from a first position to a second position], for selecting the mixing, i.e. for regulating the dilution ratio of the chemical/water mixture) and adjusting a chemical inlet back pressure by selecting a metering tip having a first orifice size (See col. 5, lines 1-7 – metering tip (not shown) with an orifice inserted into the chemical port 94 for controlling the dilution ratio of the chemical fluid), where the pressure regulating valve functions for at least one of opening and blocking the motive source, e.g. the pressurized water, but fail to teach two different valves connected in series. None of the other references identified above teach the aforementioned claimed limitations of claims 40-41 nor would it have been obvious to combine references to achieve the claimed inventive subject matter

***Response to Arguments***

24. Applicant fails to properly respond to the claim objection presented by Examiner in the non-final office action presented 11 January 2005. Accordingly, Examiner maintains the claim objection as described above in paragraph 1.

25. Applicant's arguments filed 14 September 2005, with regard to independent claims 1, 10, 19, and 27, have been fully considered but they are not persuasive. Examiner appreciates Applicant's explanation of the claimed invention during the telephone interview conducted 7 September 2005 in assisting Examiner with evaluation of Applicant's claimed invention in light of the referenced prior art references. Examiner disagrees with Applicant's Remarks/Arguments in view of the claims as presently recited. Examiner disagrees with Applicant's argument that the ball valve, as taught by Sand et. al., merely functions to turn the water on and off, i.e. acting like a light switch to open or close the motive fluid path only. As presently recited, Examiner maintains that Sand et. al. teach the method comprising the step of hooking a water supply source to the water inlet port of the eductor (See col. 4, lines 19-20 – chemical fluids diluted to a desired dilution ratio by mixing with a motive fluid of pressurized water; lines 24-29 – motive fluid is received at an inlet valve 34, depicted as a ball valve, and passed through a motive fluid conduit 36 to the device 32 for mixing a selected chemical fluid, drawn respectively from chemical fluid reservoirs; and lines 39-40 – device 32 includes a single selection control, depicted as a lever 52 [for activating the ball valve from a first position to a second position], for selecting the mixing, i.e. for regulating the dilution ratio of the chemical/water mixture. Accordingly, the ball valve [inlet valve 34] operated by lever 52 functions a pressure regulating valve for regulating a working pressure of the water supply (pressurized water – See col. 4, line 20) from a first pressure to a second pressure (adjusting lever 52 from a first position with a first

pressure to a second position with a second pressure), and regulating the pressure regulating valve (the ball valve/inlet valve 34) based on a first pressure (completely closed or at a first position with a first pressure) to maintain the water supply source at substantially a second pressure (completely open or at a operator-determined second position with a second position) to produce a regulated working pressure of the water supply for controlling the desired dilution ratio of the chemical fluids with the motive fluid. Applicant further argues that Sand et. al. does not disclose both a pressure regulating valve to regulate water supply pressure and a separate valve to activate the eductor. Examiner has previously identified that Sand et. al. does disclose a pressure regulating valve to regulate water supply pressure to the eductor. Examiner agrees that a separate valve is not utilized to activate the eductor; however, Applicant's presently recited claims do not teach that the step of activating the eductor "by opening a valve" requires that a second valve be present. Accordingly, Sand et. al. teach the use of lever 52 and inlet ball valve 34 and/or a selection member to activate the eductor by opening the valve to mix [pressurized] water and chemical fluids to a desired dilution ratio containing a volume of sterilant to a volume of water (See col. 4, lines 24-28 and 39-40; See col. 6, lines 46-54 – a dispenser 32 mixes a selected chemical fluid when an operator positions a selection member, such as a selector disk 68, concentric disk selection member 122, push button selection member 152, and a cylindrical section member 202. Thereby, the device 32 diverts motive fluid to one or more fluids channels, such as to the first eductor 92 that draws a first chemical fluid for mixing, to the second eductor 96 that draws a second chemical fluid for mixing, or to the motive fluid bypass 100). Accordingly, Examiner maintains all rejections to the presently amended and original claims, as described above in paragraphs 2-22.

### ***Conclusion***

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26. Applicant's amendment necessitated the new ground(s) of rejection presented in this Office action. Accordingly, **THIS ACTION IS MADE FINAL**. See MPEP § 706.07(a). Applicant is reminded of the extension of time policy as set forth in 37 CFR 1.136(a).

A shortened statutory period for reply to this final action is set to expire THREE MONTHS from the mailing date of this action. In the event a first reply is filed within TWO MONTHS of the mailing date of this final action and the advisory action is not mailed until after the end of the THREE-MONTH shortened statutory period, then the shortened statutory period will expire on the date the advisory action is mailed, and any extension fee pursuant to 37 CFR 1.136(a) will be calculated from the mailing date of the advisory action. In no event, however, will the statutory period for reply expire later than SIX MONTHS from the date of this final action.

Any inquiry concerning this communication or earlier communications from the examiner should be directed to Brad Y. Chin whose telephone number is 571-272-2071. The examiner can normally be reached on Monday – Friday, 8:00 A.M. – 5:00 P.M.

If attempts to reach the examiner by telephone are unsuccessful, the examiner's supervisor, Sun (John) Kim, can be reached at 571-272-1142. The fax phone number for the organization where this application or proceeding is assigned is 703-872-9306.

Information regarding the status of an application may be obtained from the Patent Application Information Retrieval (PAIR) system. Status information for published applications may be obtained from either Private PAIR or Public PAIR. Status information for unpublished applications is available through Private PAIR only. For more information about the PAIR system, see <http://pair-direct.uspto.gov>. Should you have questions on access to the Private PAIR system, contact the Electronic Business Center (EBC) at 866-217-9197 (toll-free):

byc

October 24, 2005

*John Kim*  
**JOHN KIM**  
**SUPERVISORY PATENT EXAMINER**